Topic 2 Molecular Biology

Skills:

- 2.1.S1 Drawing molecular diagrams of glucose, ribose, a saturated fatty acid and a generalized amino acid.
- 2.1.S2 Identification of biochemicals such as sugars, lipids or amino acids from molecular diagrams.

- A. Basics of drawing diagrams
 - 1. Write out the chemical formula if it not given.
 - 2. Draw the carbons first (the carbon backbone of the molecule).
 - 3. Add functional groups that are found as a part of the molecule.
 - 4. Review over the structure to ensure that all of the different atom types are showing the correct number of covalent bonds for that type of element.
 - 5. Check the structure against the chemical formula by counting the numbers of each type of atom.

- B. Covalent bonds between atoms are as follows:
 - 1. Single 2 pairs of electrons shared: 1 line
 - 2. Double 4 pairs of electrons shared: 2 lines
 - 3. Triple 6 pairs of electrons shared: 3 lines
- C. All atoms must have 8 valence electrons through bonding except for hydrogen which must have 2 valence electrons.
 - 1. Carbon will form single, double, and triple bonds, with combinations of the three types of bonds.
 - 2. Hydrogen only will have 1 single bond.
 - 3. Oxygen will often have 2 single or 1 double bond.
 - 4. Nitrogen will usually have 3 single bonds, but may have any combination of single, double and or triple bonds.

C. Memorizing functional groups make drawing easier:

$$-OH$$

$$-NH_2$$

or

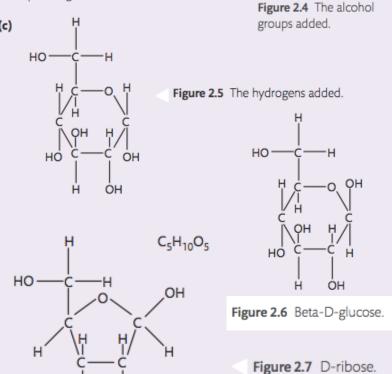
H

Η

or

D. Carbohydrates:

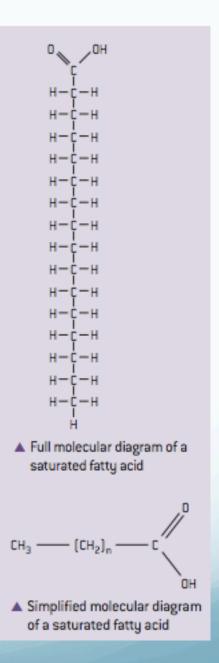
- The carbon backbone in water form rings instead of a long chain.
- 2. The top most portion of the ring will be an oxygen atom, to the right is the alpha (α) carbon or carbon 1 (C_1).
- 3. Continue drawing carbons as a pentagon for ribose (a pentose sugar) or a hexagon for glucose (a hexose sugar).
- The last carbon will not be included in the ring, but placed above the last carbon in the ring.
- 5. Each carbon in the ring will have one hydroxyl group and one hydrogen. The lone carbon will have a hydroxyl group and two hydrogen atoms. Note their positions, up and down on the ring.



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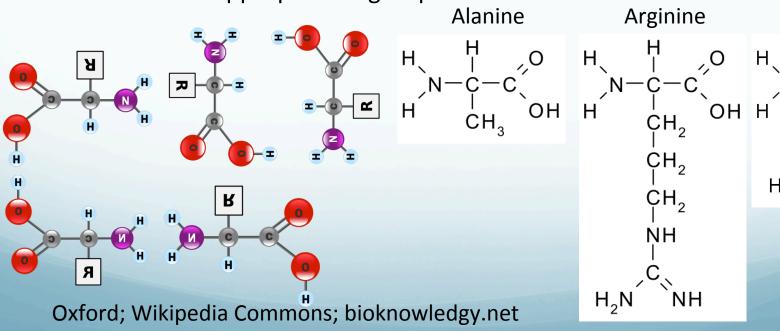
D. Saturated fatty acids:

- 1. Draw a carboxyl group at the head.
- 2. The attached carbon atoms will form an unbranched chain.
- Typically the number of carbons in the chain will be from 14 to 20 in length.
- 4. Add 2 hydrogen atoms to each carbon in the chain.
- At the end of the chain add 3 hydrogen atoms to the carbon.



E. Amino acids:

- 1. Draw a single carbon atom.
- 2. Add an amine group (amino).
- 3. Add a carboxyl group (an acid) opposite the amine group.
- 4. Add a hydrogen atom.
- 5. Add the appropriate R group.



Leucine

Which molecule represents ribose?

Past paper question

A.

В.

$$\begin{array}{c|c} H & \begin{array}{c} O \\ \parallel \\ \end{array} \\ H & \begin{array}{c} M \end{array} \\ \end{array}$$

What is molecule B?

C.

$$H_2$$
C — OH

 H C — OH

 H_2 C — OH

D.

What is molecule C?

Which structure represents an amino acid?

A.

B.

C

$$\begin{array}{c|c}
H & H & O \\
N & C & C \\
H & OH & H
\end{array}$$

D.

$$CH_3$$
 \longrightarrow $(CH_2)_n$ \longrightarrow C OH_3

H

Past paper question

Which molecule is:

- i. ribose
- ii. generalised fatty acid
- iii. generalised amino acid

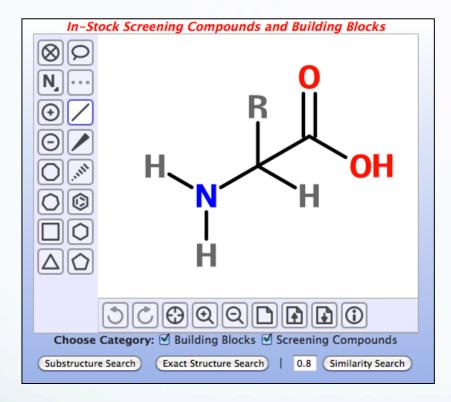
III.

IV.

$$H$$
 N
 C
 C
 OH

Discuss which two molecules are most similar in structure.

Practice



Try drawing by hand (or on eMolecules, ChemSpider, etc.) the following molecules:

- A. alpha D and beta D Glucose
- B. Ribose
- C. A generalized saturated fatty acid
- D. A generalized amino acid
- E. An example of an amino acid e.g. Alanine (the simplest)

You need to test yourself.

- Practice drawing all 5.
- Then draw all 5 in 15 minutes.
- Come back and try it again one week later to make sure you remember.